His···Asp Catalytic Dyad of Ribonuclease A: Structure and Function of the Wild-Type, D121N, and D121A Enzymes[†]

L. Wayne Schultz, David J. Quirk,‡ and Ronald T. Raines*

Departments of Biochemistry and Chemistry, University of Wisconsin—Madison, Madison, Wisconsin 53706

Received November 10, 1997; Revised Manuscript Received April 17, 1998

ABSTRACT: The side chains of histidine and aspartate residues form a hydrogen bond in the active sites of many enzymes. In serine proteases, the His...Asp hydrogen bond of the catalytic triad is known to contribute greatly to catalysis, perhaps via the formation of a low-barrier hydrogen bond. In bovine pancreatic ribonuclease A (RNase A), the His...Asp dyad is composed of His119 and Asp121. Previously, site-directed mutagenesis was used to show that His119 has a fundamental role, to act as an acid during catalysis of RNA cleavage [Thompson, J. E., and Raines, R. T. (1994) J. Am. Chem. Soc. 116, 5467-5468]. Here, Asp121 was replaced with an asparagine or alanine residue. The crystalline structures of the two variants were determined by X-ray diffraction analysis to a resolution of 1.6 Å with an R-factor of 0.18. Replacing Asp121 with an asparagine or alanine residue does not perturb the overall conformation of the enzyme. In the structure of D121N RNase A, N_{δ} rather than O_{δ} of Asn121 faces His119. This alignment in the crystalline state is unlikely to exist in solution because catalysis by the D121N variant is not compromised severely. The steady-state kinetic parameters for catalysis by the wild-type and variant enzymes were determined for the cleavage of uridylyl(3'→5')adenosine and poly(cytidylic acid), and for the hydrolysis of uridine 2',3'-cyclic phosphate. Replacing Asp121 decreases the values of k_{cat}/K_{m} and k_{cat} for cleavage by 10-fold (D121N) and 10²-fold (D121A). Replacing Asp121 also decreases the values of $k_{\text{cat}}/K_{\text{m}}$ and k_{cat} for hydrolysis by 10^{0.5}-fold (D121N) and 10-fold (D121A) but has no other effect on the pH-rate profiles for hydrolysis. There is no evidence for the formation of a low-barrier hydrogen bond between His119 and either an aspartate or an asparagine residue at position 121. Apparently, the major role of Asp121 is to orient the proper tautomer of His119 for catalysis. Thus, the mere presence of a His...Asp dyad in an enzymic active site is not a mandate for its being crucial in effecting catalysis.

The amino acid motifs available to enzymic catalysts are diverse. Still, many enzymes fall into classes wherein great similarities exist. One well-studied class is that of the serine proteases (1). This class of enzymes has an active-site motif known as the catalytic triad, which is composed of the residues Ser···His···Asp linked by hydrogen bonds (2, 3).

The importance ascribed to the Ser···His···Asp motif is in large part due to its presence in the active sites of nonhomologous serine proteases (4). In recent years, the triad has also been found in several lipases (5, 6), Fusarium solani cutinase (7), and the Sindbis virus core protein (8). Replacing the aspartate residue in the catalytic triad of human lipoprotein lipase with a glycine is responsible for familial type I hyperlipoproteinemia (9).

Similar to the catalytic triad, a His···Asp motif is found in the active sites of serine carboxypeptidase (10), acetylcholinesterase (11), phospholipase A₂ (12), haloalkane dehalogenase (13), dienelactone hydrolase (14), a variety of zinc-dependent enzymes (15), and pancreatic ribonucleases

(16, 17). This His...Asp motif is known as the "catalytic dyad". With the exception of the pancreatic ribonucleases, enzymes containing the His...Asp dyad have syn-oriented carboxylates (18), though the benefit to catalysis of a syn versus anti orientation appears to be minimal (19, 20).

In all enzymes containing a catalytic triad or dyad, a hydroxyl group functions as a nucleophile. The role of the histidine residue is to abstract a proton from the hydroxyl group and thereby to increase the nucleophilicity of the oxygen. The role of the aspartate residue has been the subject of much study and debate, as has the importance of the catalytic triad in general (21).

The catalytic triad was once thought to act in a "charge relay mechanism". Specifically, it was postulated that a proton was removed from the serine residue by histidine while another proton was being removed from histidine by aspartate (2, 22). The unperturbed pK_a of aspartic acid is too low to favor such a mechanism, and the general consensus today is that the charge relay mechanism is not operative. Instead, the aspartate residue forms a hydrogen bond that could serve both to orient the histidine side chain and to increase its pK_a , enhancing its ability to abstract a proton from the serine residue.

Site-directed mutagenesis has been used to illuminate the role and importance of the aspartate residue in the catalytic triad of two serine proteases. Replacing the aspartate residue

[†] This work was supported by Grant GM44783 (NIH). X-ray data collection and computational facilities were supported by Grant BIR-9317398 (NSF). L.W.S. was supported by Postdoctoral Fellowship CA69750 (NIH). D.J.Q. was supported by Cellular and Molecular Biology Training Grant GM07215 (NIH).

^{*} To whom correspondence should be addressed.

[‡] Present address: School of Pharmacy, University of Wisconsin—Madison, Madison, WI 53706.

FIGURE 1: Putative mechanism for the transphosphorylation (A) and hydrolysis (B) reactions catalyzed by RNase A (33). In the hydrolysis reaction, the hydrogen bonds in Asp $121\cdots$ His $119\cdots$ H₂O resemble those in the renown catalytic triad of serine proteases.

with asparagine in trypsin (23, 24) or alanine in subtilisin (25) leads to a 10⁴-fold reduction in catalytic activity. This decrease suggests that the aspartate residue plays a critical role in catalysis. X-ray diffraction analysis of the trypsin variant revealed, however, that the catalytic histidine residue was being stabilized as an unproductive tautomer by its interaction with asparagine (26). This result suggests that one role of the aspartate residue in a Ser···His···Asp triad is to orient the proper tautomer of histidine.

Two attempts have thus far been made to determine the role of Asp121, the aspartate in the catalytic dyad of bovine pancreatic ribonuclease A [RNase A,¹ EC 3.1.27.5 (27)]. In one study, Asp121 was replaced with asparagine in a semisynthetic enzyme (28). This semisynthetic RNase A, RNase(1-118)•(111-124), consists of a noncovalent complex between residues 1-118 of RNase A (obtained from proteolytic digestion of RNase A) and an overlapping synthetic peptide composed of the 14 C-terminal residues of RNase A, except with Asp121 replaced by an asparagine residue. The semisynthetic D121N analogue has 5% of the catalytic activity of the analogous wild-type semisynthetic enzyme. A difficulty with interpreting the results of this

study, however, is that the three-dimensional structure of the semisynthetic D121N analogue exhibits numerous changes compared to that of the analogous wild-type semisynthetic enzyme (29, 30). In another study, site-directed mutagenesis was used to replace Asp121 in RNase A itself with a glutamate residue (31). The D121E enzyme has 17% of the activity of the wild-type enzyme for the hydrolysis of 0.1 mM C>p and 16% for the hydrolysis of 0.4 mM C>p. No other characterization of this variant has been reported. In related work, replacing Asp116 of the His···Asp dyad of angiogenin, a homologue of RNase A, with an asparagine or alanine residue was shown to *increase* ribonucleolytic activity by 8- and 15-fold, respectively (32).

Here, we address the role of the aspartate residue in the catalytic dyad of RNase A. RNase A catalyzes the twostep hydrolysis of the P-O5' bond of RNA on the 3' side of pyrimidine residues. Figure 1 depicts the classical mechanism for these reactions (33). In the transphosphorylation step, His119 of the His···Asp dyad acts as an acid. The slow hydrolysis of the 2',3'-cyclic phosphate occurs separately and resembles the reverse of transphosphorylation (34, 35). It is in this second step that the action of the catalytic dyad of RNase A most resembles that of the catalytic triad. The amino acid sequences of pancreatic ribonucleases from more than 40 vertebrates are known (16, 17) and are evolving rapidly (36). The catalytic dyad is conserved in each of these ribonucleases, suggesting that these residues are important. We have created RNase A variants in which Asp121 is replaced with an asparagine or alanine residue. Here, we report on the three-dimensional structures of the variants, as well as on their abilities to catalyze transphosphorylation and hydrolysis.

EXPERIMENTAL PROCEDURES

Materials. Escherichia coli strain BL21(DE3) (F⁻ ompT r_B-m_B-) (*37*) was from Novagen (Madison, WI). Buffers (except Tris), cUMP, and IPTG were from Sigma Chemical (St. Louis, MO). Tris was from Fisher Scientific (Pittsburgh, PA). Poly(C) was from Midland Reagent (Midland, TX).

¹ Abbreviations: AMPSO, 3-[(1,1-dimethyl-2-hydroxyethyl)amino]-2-hydroxypropanesulfonic acid; BICINE, N,N-bis(2-hydroxyethyl)glycine; Bis-Tris, bis(2-hydroxyethyl)iminotris(hydroxymethyl)methane; BSA, bovine serum albumin; cCMP, cytidine 2',3'-cyclic phosphate (otherwise C>p); cUMP, uridine 2',3'-cyclic phosphate (otherwise U>p); DEAE, diethylaminoethyl; EDTA, ethylenediaminetetraacetic acid; HEPES, N-(2-hydroxyethyl)piperazine-N'-2-ethanesulfonic acid; IPTG, isopropyl β -D-thiogalactopyranoside; MES, 2-(N-morpholino)ethanesulfonic acid; MOPS, 3-(N-morpholino)propanesulfonic acid; MPD, 2-methyl-2,4-pentanediol; OD, optical density; poly(C), poly-(cytidylic acid); PAGE, polyacrylamide gel electrophoresis; PDB, Protein Data Bank, which is maintained by the Brookhaven National Laboratory; ppm, parts per million; RNase A, bovine pancreatic ribonuclease A; sD121A RNase A, semisynthetic RNase A consisting of a noncovalent complex between residues 1-118 and residues 111-124, with Asp121 replaced by an alanine residue; sD121N RNase A, semisynthetic RNase A consisting of a noncovalent complex between residues 1-118 and residues 111-124, with Asp121 replaced by an asparagine residue; SDS, sodium dodecyl sulfate; TB, terrific broth; Tris, tris(hydroxymethyl)aminomethane; 3'-UMP, uridine 3'-phosphate (otherwise Up); UpA, uridylyl(3'→5')adenosine.

UpA was synthesized by J. E. Thompson using methods published previously (38, 39). Growth media were from Difco (Detroit, MI) or, for 10 L growths, Marcor Development (Hackensack, NJ). DNA sequences were determined with the Sequenase Version 2.0 kit from U.S. Biochemicals (Cleveland, OH). DEAE Sephadex A-25 anion exchange resin, S-Sepharose cation exchange resin, and the mono-S FPLC cation exchange column were from Pharmacia LKB (Piscataway, NJ). Bacterial terrific broth [TB (40)] contained (in 1 L) tryptone (12 g), yeast extract (24 g), glycerol (4 mL), KH₂PO₄ (2.3 g), and K₂HPO₄ (12.5 g).

Mutagenesis. Previously, we described the construction of a bovine pancreatic cDNA library, the cloning of the cDNA that encodes RNase A, and the efficient expression of this cDNA in E. coli (41). Here, the codon for Asp121 was changed to one for an asparagine or alanine residue by oligonucleotide-mediated site-directed mutagenesis (42) using the oligonucleotides 5'-ACTACTACACGCTAGCGTT-AAAGTGGACT-3' and 5'-ACTACTACACGCTAGCGGC-AAAGTGGACT-3', where the underlined bases differ from those in the wild-type cDNA. The GCT mismatch introduced a unique, translationally silent NheI site that was used to screen for mutated plasmids. The complete sequence of the cDNA that encodes each enzyme was determined. Plasmids that direct the expression of wild-type, D121N, and D121A RNase A were transformed into E. coli BL21(DE3).

Production of RNase A in E. coli. TB (0.20 L) containing ampicillin (200 μ g/mL) was inoculated with a culture frozen in aqueous glycerol (30% w/v). After growth overnight at 37 °C, cells were collected by centrifugation and suspended in 0.20 L of fresh TB. This suspension was used to inoculate 10 L of TB containing ampicillin (200 μ g/mL). When OD = 4 at 600 nm, IPTG was added to a final concentration of 0.5 mM. The culture was grown for an additional 3.5 h.

Cells were collected by centrifugation, resuspended in 0.10 L of cold TE buffer, and lysed by passage through a French pressure cell twice. The lysate was centrifuged at 30000g for 30 min. The resulting pellet was washed with a cold solution (0.30 L) of deoxycholate (1 mg/mL) and centrifuged again. The pellet was suspended in 0.50 L of Tris-acetic acid buffer (pH 8.0) containing urea (9 M), acetic acid (50 mM), sarcosine (40 mM), and EDTA (1 mM). Sonication was used to aid dissolution. To reduce any disulfide bonds, dithiothreitol was added to a final concentration of 50 mM. After 1 h, the solution was passed through DEAE A-25 resin (25 g) that had been equilibrated with the above resuspension solution, and acetic acid ($\frac{1}{10}$ volume) was added to the effluent. The resulting solution was dialyzed exhaustively against 20 mM acetic acid. The dialysate was centrifuged, the supernatant diluted to 1 L, and the pH raised to 8.1 by the addition of Tris base. Reduced glutathione and oxidized glutathione were then added to final concentrations of 2.3 and 0.8 mM, respectively. After 24 h, the resulting solution was loaded onto an S-Sepharose cation exchange column (0.20 L). The column was washed with 1 L of 50 mM HEPES-HCl buffer (pH 7.9), and the adsorbed proteins were eluted with a linear gradient (0.50 L + 0.50 L) of NaCl (0 to 0.50 M). Fractions were pooled on the basis of A at 280 nm, and the pooled fractions were diluted with buffer to a protein concentration of <10 mg/mL. The resulting solution was loaded onto an HR10/10 mono-S cation exchange FPLC column (15 cm \times 1.8 cm²) that had been equilibrated with 50 mM HEPES-HCl buffer (pH 7.9). The column was washed with 50 mM HEPES-HCl buffer (pH 7.9), and the adsorbed proteins were eluted with a linear gradient (50 mL + 50 mL) of NaCl (0 to 0.50 M) in the same buffer. The proteins eluted between 0.09 and 0.12 M NaCl. Fractions were again pooled on the basis of A at 280 nm, which was monitored continuously. To discourage deamination, the pH of the pooled fractions was lowered to 5.5 by the addition of 20 mM acetic acid. After exhaustive dialysis against water, the proteins were concentrated with an Amicon Centriprep-10 filter, passed through a 0.45 μ m filter (Schleicher & Schuell, Keene, NH), and stored at 4 °C.

The purity of the protein preparation was assessed by SDS-PAGE and native PAGE. On the basis of the symmetry of A_{280} peaks during cation exchange chromatography and the appearance of bands after electrophoresis through polyacrylamide gels, wild-type and the D121N variant were $\geq 99\%$ pure and the D121A variant was $\geq 95\%$ pure. The concentration of purified enzyme was determined by assuming that A=0.72 at 277.5 nm for a 1.0 mg/mL solution (43). A 10 L growth typically yielded several hundred milligrams of purified enzyme.

Crystallization. Salt-free aqueous solutions of the D121N and D121A variants were lyophilized. Lyophilized D121N RNase A was dissolved in unbuffered water to a concentration of 60 mg/mL. Crystallization was performed in small siliconized tubes using the batch method. A solution (25 μ L) of enzyme was mixed with an equal volume of 95% (v/v) 2-methyl-2,4-pentanediol (MPD) in 0.10 M MES-NaOH buffer (pH 5.2) to give a final enzyme concentration of 30 mg/mL in 47.5% (v/v) MPD and 0.05 M MES-NaOH buffer (pH 5.2). The pH of the buffer was determined prior to adding the MPD. Monoclinic crystals of D121N RNase A appeared after several months of incubation at 20 °C. These crystals belong to space group $P2_1$ with the following dimensions: a = 30.52 Å, b = 38.40 Å, c = 53.31 Å, and $\beta = 105.7$ °.

Lyophilized D121A RNase A was dissolved in unbuffered water to a concentration of 60 mg/mL. Crystallization was performed using the hanging drop method. Drops consisting of enzyme solution (1.5 μ L), water (1.5 μ L), and reservoir solution (3.0 μ L) were suspended from siliconized coverslips over 0.50 mL of 2.5 M sodium acetate buffer (pH 5.3) containing saturating NaCl. Trigonal crystals of the D121A RNase A appeared within 1 day of incubation at 20 °C and grew to a size of 0.5 mm \times 0.5 mm \times 1.0 mm after several days. These crystals belong to space group $P3_221$ with the following dimensions: a = b = 64.70 Å, c = 65.03 Å, and $\rho = 120^\circ$.

X-ray Diffraction Data Collection. All X-ray data were collected with a Siemens HI-STAR detector mounted on a Rigaku rotating anode operating at 50 kV and 90 mA and a 300 μ m focal spot. The X-ray beam was collimated by double-focusing mirrors. The crystal to detector distance was 12.0 cm. Data were obtained in 512 × 512 pixel format, processed with the program XDS (44, 45), and scaled using the program XSCALIBRE (G. Wesenberg and I. Rayment, unpublished results). The data set for D121N RNase A was 88% complete in the resolution range of 30–1.6 Å with an overall merging *R*-factor of 1.9%. The data set for D121A RNase A was 91% complete in the resolution range of 30–1.6 Å with an overall merging *R*-factor of 2.9%.

Structure Refinement. The starting model for the structure of D121N RNase A consisted of residues 2-124 of the wildtype enzyme (PDB entry 7rsa), stripped of all solvent molecules and side chain atoms of residue 121. The model was subjected to 10 cycles of least-squares refinement with the program TNT (46), yielding an initial R-factor of 23.4%. A difference Fourier map $(F_o - F_c)$ clearly showed the position of the side chain atoms of Asn121. Manual adjustments to the model were performed by using the program FRODO (47). After several cycles of manual adjustments and least-squares refinement, the resolution was extended to 1.6 Å and water molecules were added to the model. The peak searching algorithm in TNT was used to place ordered water molecules. Water molecules with at least 1σ of $2F_{\rm o} - F_{\rm c}$ density and 3σ of $F_{\rm o} - F_{\rm c}$ density and within hydrogen bonding distance of the protein or of other water molecules were retained.

The structure of the D121N variant of RNase A was solved in the monoclinic space group $P2_1$ and contains only water molecules in the active site. Here, the D121N structure will be compared to that of phosphate-free wild-type RNase A, which was also crystallized from organic solvent in the $P2_1$ space group [PDB entry 7rsa (48)]. The D121N structure will also be compared with the semisynthetic D121N RNase A analogue (sD121N), which contains a sulfate ion in the active site and was crystallized from high salt in the trigonal space group $P3_221$. The sD121N structure was solved to a resolution of 2.0 Å [PDB entry 3srn (30)]. In a least-squares superposition, the main chain atoms of D121N RNase A have an average rms deviation of 0.15 Å from those of the wildtype enzyme and 0.75 Å from the sD121N analogue.

The starting model for the structure of D121A RNase A consisted of residues 1-124 of wild-type RNase A (PDB) entry 1rph), stripped of all solvent molecules and side chain atoms of residue 121. The model was subjected to 10 cycles of least-squares refinement with the program TNT (46), yielding an initial R-factor of 22.5%. A difference Fourier map $(F_{\rm o} - F_{\rm c})$ clearly showed the position of the side chain carbon of Ala121. Manual adjustments to the model were performed by using the program FRODO (47). After several cycles of manual adjustments and least-squares refinement, the resolution was extended to 1.6 Å and water molecules were added to the model. The peak searching algorithm in TNT was used to place ordered water molecules. Water molecules with at least $1\sigma 2F_o - F_c$ density and $3\sigma F_o - F_c$ density and within hydrogen bonding distance of the protein or of other water molecules were retained.

The structure of the D121A variant of RNase A was solved in the trigonal space group $P3_221$ and contains the first example of an acetate ion in an RNase A active site. Here, the D121A structure will be compared to that of wild-type RNase A that was also crystallized from high salt in the P32-21 space group and contains a bound sulfate in the active site [PDB entry 1rph (49)]. The D121A structure will also be compared with that of the semisynthetic D121A RNase A analogue (sD121A), which contains a sulfate ion in the active site and was crystallized from high salt in the P3₂21 space group. The sD121A structure was solved to a resolution of 2.0 Å [PDB entry 4srn (30)]. In a least-squares superposition, the main chain atoms of D121A RNase A have an average rms deviation of 0.38 Å from those of the wildtype enzyme and 0.53 Å from the sD121A analogue.

Enzyme Kinetics. Steady-state kinetic parameters for the cleavage of UpA and poly(C) and the hydrolysis of cUMP were determined by spectrophotometric assays. The catalytic dyad of RNase A most resembles the well-characterized catalytic triad of serine proteases during the hydrolysis reaction (Figure 1). For this reason, the steady-state kinetic parameters for the hydrolysis reaction were evaluated as a function of pH.

Assays were performed at 25 °C in a 0.10 M solution of the appropriate buffer, adjusted in pH with aqueous NaOH or HCl. Enough NaCl was added to each reaction to increase the ionic strength I to 0.10 M, which was calculated on the basis of the concentrations of buffer components and substrate. The buffer used for assays of the transphosphorylation of UpA (0.10-1.0 mM) and poly(C) (0.050-1.0 mM) was MES (pH 6.0). The buffers used for assays of the hydrolysis of cUMP were formic acid (pH 3.45 and 4.00), acetic acid (pH 4.48 and 5.02), MES (pH 5.53 and 6.03), Bis-Tris (pH 6.00 and 7.02), BICINE (pH 6.48, 7.97, and 8.46), MOPS (pH 6.94 and 7.46), and AMPSO (pH 9.00).

Assays were performed in a 1.0 or 0.2 cm cuvette with a Cary model 3 spectrophotometer equipped with a Cary temperature controller. Substrate concentrations were determined by ultraviolet absorption using an ϵ_{260} of 24 600 M^{-1} cm⁻¹ at pH 7.0 for UpA (50) and an ϵ_{268} of 6200 M^{-1} cm^{-1} at pH 7.8 for poly(C) (51) or by mass for cUMP. The concentrations of substrate in the assays ranged from $K_{\rm m}/5$ to $5K_{\rm m}$. BSA (0.5 mg/mL) was added to the reaction mixtures containing poly(C). The cleavage of UpA was monitored at 286 nm using a $\Delta\epsilon$ of $-755 \text{ M}^{-1} \text{ cm}^{-1}$. The cleavage of poly(C) was monitored at 238 nm. The $\Delta\epsilon$ at 250 nm, as calculated from the difference in molar absorptivity of the polymeric substrate and the mononucleotide cyclic phosphate product, has been reported to be 2380 M⁻¹ cm⁻¹ at pH 6.2 (51). The $\Delta\epsilon$ at 238 nm was determined to be 2792 M⁻¹ cm⁻¹ by observing the change in absorption of a partially cleaved substrate at 250 and 238 nm. The hydrolysis of cUMP was monitored at either 282 or 293 nm. The values for the extinction coefficients determined for the hydrolysis of cUMP are listed in Table 3.

Kinetic Data Analysis. Data processing and nonlinear regression analysis were performed with the program MATH-CAD (MathSoft, Cambridge, MA). Initial velocity data were weighted in proportion to the magnitude of each data point. The pH-rate profiles were fitted by using the logarithm of the kinetic constants. Reported errors were generated by the asymptotic variance—covariance matrix method. Because of strong product inhibition and a relatively small change in the extinction coefficient, the initial velocities for the hydrolysis of cUMP were difficult to measure reliably. For this reason, the steady-state kinetic parameters for this substrate were determined by using the integrated rate equation (52, 53) to fit the complete time course of each hydrolysis reaction. The use of the integrated rate equation provided more accurate values for the steady-state kinetic parameters as well as values for K_p , the dissociation constant for the RNase A·3'-UMP complex. Each progress curve for the hydrolysis reaction was fitted to the integrated rate equation with the inclusion of product inhibition by an iterative procedure in which the kinetic parameters were varied by hand and the fit was judged by eye. The value of $k_{\rm cat}$ was allowed to vary slightly from progress curve to

Table 1: X-ray Diffraction Analysis Statistics crystal data D121N RNase A D121A RNase A $P2_{1}$ $P3_{2}21$ space group a = 30.52 Å (1)a = 64.70 Å (1)cell dimensions b = 38.40 Å (1)b = 64.70 Å (1)c = 53.31 Å (1)c = 65.03 Å (2) $\beta = 105.7^{\circ} (1)$ protein molecules/unit cell **Data Collection Statistics** resolution (Å) 1.6 1.6 56 287 no. of measured reflections 28 480 no. of unique reflections 15 756 27 087 completeness of data (%) 89 91 $R_{\text{merge}} (I/\sigma > 0.33)^a$ 0.019 0.029 average I/σ (1.7–1.6 Å) 4.5 1.9 $R_{\text{merge}} (1.7 - 1.6 \text{ Å})$ 0.055 0.200 Final Refinement Statistics 951 RNase A atoms 954 solvent atoms 102 103 R-factor $(30.0-1.6 \text{ Å})^b$ 0.177 0.180 rms deviation from ideal geometry 0.01 bond distances (Å) 0.01bond angles (deg) 2.1 2.7 average B-factors (Å²) protein (main chain) 16.5 23.8 28.4 22.1 protein (all atoms) acetate 29.7 39.0 41.4 solvent

 $^aR_{\mathrm{merge}} = \sum_{hkl} |I - \langle I \rangle| / \sum_{hkl} I$, where I = the observed intensity and $\langle I \rangle$ = the averaged intensity obtained from multiple observations of symmetry-related reflections. $^bR = \sum_{hkl} |F_o - F_c| / \sum_{hkl} |F_o|$, where F_o and F_c are the observed and calculated structure factors, respectively.

progress curve to allow for small errors in enzyme concentration, with the values of $K_{\rm m}$ and $K_{\rm p}$ being held constant. The derivative of the functions so obtained was used to determine better fitting values for the initial velocities. These velocities were then used to obtain the kinetic parameters using the Michaelis—Menten equation. If the parameters so realized differed greatly from the initial values, the process was repeated until they differed little from the prior round. Generally, one or two rounds were sufficient.

RESULTS

Oligonucleotide-mediated site-directed mutagenesis was used to change Asp121 of the His···Asp catalytic dyad of RNase A to an asparagine or alanine residue. The enzyme variants were produced in *E. coli*, purified, and crystallized. The statistics for X-ray diffraction analysis are listed in Table 1, and the structures of the active sites are shown in Figures 2 and 3.

The final model for D121N contains the complete protein (residues 1–124) and 102 water molecules. The *R*-factor for all data in the range of 30–1.6 Å is 0.177. The rms deviations for target geometries are 0.011 Å for bond lengths and 2.1° for bond angles. Average *B*-factors for the main chain and all atoms are 16.5 and 22.1 Ų, respectively. Atomic coordinates for D121N RNase A have been deposited in the Protein Data Bank under accession code 3rsd.

The final model for D121A contains the complete protein (residues 1-124), 100 water molecules, two chloride ions, and an acetate ion. The current *R*-factor for all data in the range of 30-1.6 Å is 0.180. The rms deviations for target geometries are 0.013 Å for bond lengths and 2.7° for bond

angles. Average B-factors for the main chain and all atoms are 23.8 and 28.4 Å², respectively. Atomic coordinates for D121A RNase A have been deposited in the Protein Data Bank under accession code 4rsd.

The steady-state kinetic parameters for the turnover at pH 6.0 of UpA, poly(C), and cUMP by wild-type RNase A and the D121N and D121A variants are listed in Table 2. The variations primarily affect the values of $k_{\rm cat}$. For all three substrates, the $k_{\rm cat}$ for D121A RNase A is approximately 6-fold lower than that of the D121N enzyme and 20–50-fold lower than that of wild-type RNase A.

Kinetic constants for the hydrolysis of cUMP at different values of pH are listed in Table 3. These data are plotted for $k_{\text{cat}}/K_{\text{m}}$, k_{cat} , and K_{p} in Figures 4–6. The solid lines in Figures 4–6 are drawn according to the following equations:

$$\frac{K_{\text{m}}}{(1 + [\text{H}^+]/K_{\text{a}})(1 + K_{\text{b}}/[\text{H}^+])(1 + [\text{H}^+]/K_{\text{c}})(1 + K_{\text{d}}/[\text{H}^+])}$$
(1)

$$k_{\text{cat}} = \frac{k_{\text{cat}}^{\text{int2}}}{1 + K_{\text{c,s}}/[\text{H}^+]} + \frac{k_{\text{cat}}^{\text{int}}}{1 + [\text{H}^+]/K_{\text{c,s}}}$$

$$(2)$$

$$K_{\rm p} = \frac{K_{\rm p}^{\rm int}(1 + K_{\rm a}/[{\rm H}^+])(1 + K_{\rm b}/[{\rm H}^+])(1 + [{\rm H}^+]/K_{\rm p})}{(1 + K_{\rm a,p}/[{\rm H}^+])(1 + K_{\rm b,p}/[{\rm H}^+])(1 + [{\rm H}^+]/K_{\rm p,p})}$$
(3)

where [H⁺] is the hydrogen ion concentration; $k_{\rm cat}/K_{\rm m}^{\rm int}$, $k_{\rm cat}^{\rm int}$, $k_{\rm cat}^{\rm int}$, $k_{\rm cat}^{\rm int}$, and $K_{\rm p}^{\rm int}$ are pH-independent constants; $K_{\rm a}$, $K_{\rm b}$, $K_{\rm c}$, and $K_{\rm d}$ are the acid dissociation constants of uncomplexed enzyme; $K_{\rm p}$ is the acid dissociation constant of uncomplexed 3'-UMP; and those constants with an s or p subscript are for the enzyme•substrate or enzyme•product complex (52, 53). In eq 1, the terms associated with $K_{\rm c}$ and $K_{\rm d}$ were not included in the fit when there was no indication of their presence. As no indication for $pK_{\rm p,p}$ was present, it was set at 2.0 (54). The value of $pK_{\rm p}$ was set at 5.8 (55). The free enzyme values of $K_{\rm p}$ were obtained from the $k_{\rm cat}/K_{\rm m}$ profiles. The $pK_{\rm a}$ values and intrinsic parameters calculated from eqs 1–3 are listed in Table 4.

As shown in Figure 4, wild-type RNase A, D121N RNase A, and D121A RNase A have virtually identical $k_{\rm cat}/K_{\rm m}$ profiles except for an alkaline dip in the profile of the D121N enzyme. Both wild-type RNase A and D121N RNase A display a second low-pH limb at pH 4.

As shown in Figure 5, wild-type RNase A, D121N RNase A, and D121A RNase A have similar $k_{\rm cat}$ profiles at pH <7, though that of the D121N enzyme is somewhat shallower than that of the others. At pH >7, the $k_{\rm cat}$ profile of the wild-type enzyme reaches a plateau, but those of the variants decrease. The $k_{\rm cat}$ profiles for wild-type RNase A and D121N RNase A have a slope of 1 at pH <4.5; all three enzymes have a slope of <1 between pH 4.5 and 7.0.

As shown in Figure 6, the K_p profile of D121N RNase A is similar to that of the wild-type enzyme except for a

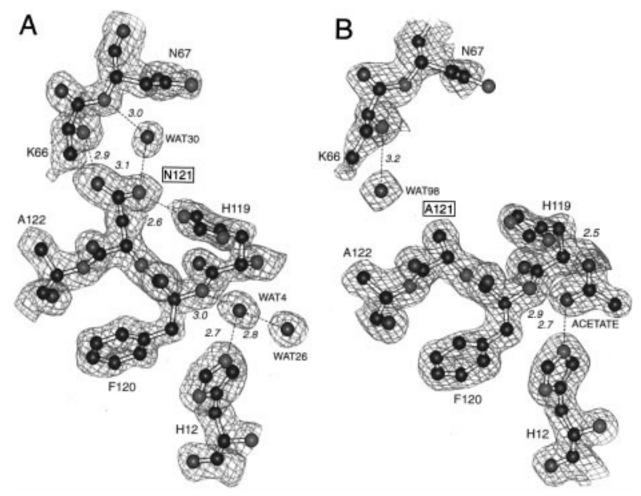


FIGURE 2: Structure of the active site of crystalline D121N RNase A (A) and D121A RNase A (B). Structures were determined by X-ray diffraction analysis to a resolution of 1.6 Å. Hydrogen bonds of ≤3.2 Å are labeled. Enzymes were crystallized at pH 5.2 (D121N RNase A) or 5.3 (D121A RNase A).

decreased K_p at high pH. The K_p of D121A RNase A is 2-fold smaller than that of the other two enzymes.

DISCUSSION

Three-Dimensional Structures. RNase A was the third enzyme (after lysozyme and carboxypeptidase) whose structure was solved by X-ray diffraction analysis (56). Despite this early precedent (57), the crystalline structures reported here are the first for active-site variants. These structures show that replacing residue Asp121 of the His···Asp catalytic dyad with an asparagine or alanine residue does not affect the structure of RNase A beyond residue 121. Interpretations of function based on the identity of the active-site residues alone are thus justified.

The structure of RNase A is comprised of a twisted fourstranded, antiparallel β -sheet consisting of two long central β -strands, β 2 and β 3 (residues 71–92 and 94–110, respectively), flanked by two shorter β -strands, $\beta 1$ and $\beta 4$ (residues 41-48 and 118-123, respectively). Three helicies are found in RNase A: $\alpha 1$ at the N terminus (residues 3–13) and $\alpha 2$ and $\alpha 3$ nearly perpendicular to and at either end of the β -sheet (residues 24–34 and 50–60, respectively). Nucleotide binding occurs in a deep cleft created by $\alpha 1$ and $\beta 3$ (58). The bases of adjacent RNA nucleotides bind in three enzymic subsites referred to as B1-3 (59). Catalysis occurs in the P1 subsite and results in the cleavage of the P-O5'

bond specifically on the 3'-side of a pyrimidine nucleotide bound in the B1 subsite. Residues from the B1 subsite and P1 catalytic site compose the walls and floor of the groove. Gln11, Lys41, and Asn44 form one wall, and His119, Asp121, and Ala122 form the other. His12, Asn44, Thr45, and Phe120 form the floor of the B1 subsite. This discussion will focus on structural deviations observed in residues constituting the active site (Figures 2 and 3).

Gln11. The side chain $N_{\epsilon 2}$ of Gln11 can donate a hydrogen bond to a phosphoryl oxygen in the active site. Site-directed mutagenesis studies in which Gln11 is replaced by an alanine, asparagine, and histidine residue have shown that the role of Gln11 is to prevent the nonproductive binding of substrate (41). In the structure of the RNase A uridine 2',3'-cyclic vanadate complex, $N_{\epsilon 2}$ of Gln11 forms a hydrogen bond with a nonbridging vanadyl oxygen, perhaps mimicking its interaction with the transition state (58, 60). The side chain $N_{\epsilon 2}$ of Gln11 forms a hydrogen bond with a bridging phosphoryl oxygen in an RNase A·d(CpA) complex and with a water molecule in an RNase A·3'-CMP complex (49).

In the wild-type monoclinic structure, $N_{\epsilon 2}$ of Gln11 forms a 3.0 Å hydrogen bond with a conserved water molecule in the active site, but $O_{\epsilon 1}$ forms no hydrogen bonds to surrounding water molecules (48). In the structure of D121N RNase A, the side chain of Gln11 adopts a slightly different

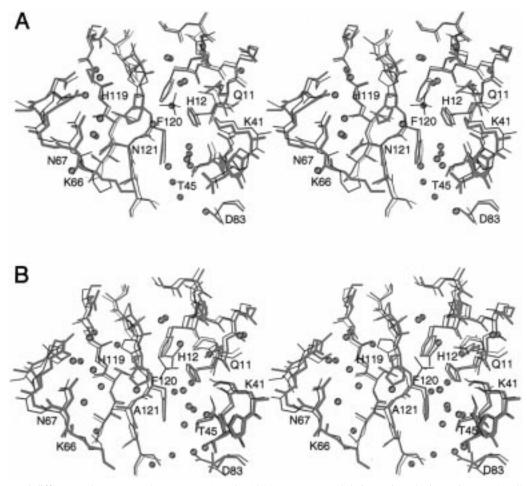


FIGURE 3: Structural differences between D121N RNase A and D121A RNase A and their semisynthetic analogues. (A) Stereoview of the overlap of the active sites of D121N RNase A (thick red lines) and the semisynthetic D121N analogue [thin blue lines (30)]. The sD121N analogue contains a sulfate ion in its active site. (B) Stereoview of the overlap of the active sites of D121A RNase A (thick red lines) and the sD121A analogue [thin blue lines (30)]. D121A RNase A contains an acetate ion in its active site, and the sD121A analogue contains a sulfate ion in its active site. Water molecules are depicted for the variants (red closed spheres) and semisynthetic analogues (blue open spheres). This figure was created with the program MOLSCRIPT (93).

Table 2: Steady-State Kinetic Parameters for Catalysis by Wild-Type RNase A, D121N RNase A, and D121A RNase A^a

						$k_{\rm cat}/K_{ m m}$	
substrate	RNase A	$k_{\rm cat}$ (s ⁻¹)	%	$K_{\rm m}$ (mM)	%	$(mM^{-1} s^{-1})$	%
UpA	wild-type	890 ± 30	100	0.38 ± 0.03	100	2300 ± 190	100
	D121N	190 ± 20	21	1.6 ± 0.2	420	120 ± 20	5
	D121A	30 ± 3	3.4	2.2 ± 0.3	580	14 ± 2	0.6
poly(C)	wild-type	475 ± 10	100	0.14 ± 0.01	100	3300 ± 200	100
	D121N	51 ± 2	11	0.10 ± 0.01	71	510 ± 60	15
	D121A	8.2 ± 0.3	1.7	0.20 ± 0.02	140	40 ± 4	1.2
cUMP	wild-type	3.7 ± 0.2	100	2.3 ± 0.5	100	1.6 ± 0.3	100
	D121N	1.55 ± 0.11	42	2.7 ± 0.7	120	0.58 ± 0.10	36
	D121A	0.187 ± 0.006	5.1	1.6 ± 0.2	70	0.11 ± 0.01	7

^a Data were obtained at 25 °C in 0.10 M MES-NaOH buffer (pH 6.0) containing NaCl (to I = 0.10 M).

conformation. The side chain $N_{\epsilon 2}$ of Gln11 forms a longer (3.4 Å) hydrogen bond with the conserved active-site water molecule, and $O_{\epsilon 1}$ forms a 2.9 Å hydrogen bond with a new water molecule. In the sD121N analogue, $N_{\epsilon 2}$ of Gln11 adopts the same conformation as it does in wild-type RNase A, forming a 2.8 Å hydrogen bond with the conserved active-site water molecule.

In the wild-type trigonal structure, $N_{\epsilon 2}$ of Gln11 forms a 2.6 Å hydrogen bond with the conserved active-site water and a 3.0 Å hydrogen bond with a sulfate oxygen bound in the active site (49). The conserved active-site water is displaced by C_1 of an acetate ion in the D121A variant. The

result is a small shift of $N_{\epsilon 2}$ away from the acetate (due to a 30° rotation around χ_3), with $O_{\epsilon 1}$ forming a hydrogen bond with a new water molecule. In the sD121A structure, $N_{\epsilon 2}$ of Gln11 forms a 2.8 Å hydrogen bond with the conserved active-site water molecule. But unlike in the wild-type enzyme, $N_{\epsilon 2}$ in the sD121A analogue does not interact with the sulfate bound in the active site.

His12. In the classical mechanism for RNase A catalysis, $N_{\epsilon 2}$ of His12 acts as a base to abstract a proton from the 2'-hydroxyl group of RNA and thereby facilitate its attack on the phosphorus atom (33, 61). In all structures of RNase A in the Protein Data Bank, His12 is found in the same

Table 3: pH Dependencies of the Steady-State Kinetic Parameters for the Hydrolysis of cUMP by Wild-Type RNase A, D121N RNase A, and D121A RNase A

	wild-type RNase A		D121N RNase A		D121A RNase A						
pH^a	$k_{\rm cat}({ m s}^{-1})$	$K_{\rm m}({ m mM})$	K _p (mM)	k_{cat} (s ⁻¹)	$K_{\rm m}$ (mM)	K _p (mM)	k_{cat} (s ⁻¹)	$K_{\rm m}$ (mM)	К _р (mM)	$\Delta\epsilon_{282}$ (M ⁻¹ cm ⁻¹)	$\begin{array}{c} \Delta\epsilon_{293} \\ (\text{M}^{-1}~\text{cm}^{-1}) \end{array}$
3.45 (F)	0.042 ± 0.001	3.79 ± 0.2	2.7	0.029 ± 0.001	5.8 ± 0.6	3.4				958	44.3
4.00 (F)	0.156 ± 0.003	2.16 ± 0.07	0.79	0.096 ± 0.003	3.1 ± 0.2	0.68				995	49.1
4.48 (A)	0.43 ± 0.01	1.45 ± 0.10	0.26	0.216 ± 0.006	2.3 ± 0.2	0.28	0.031 ± 0.001	1.66 ± 0.06	0.17	971	47.7
5.02 (A)	1.05 ± 0.01	1.34 ± 0.03	0.13	0.47 ± 0.02	1.6 ± 0.2	0.10	0.055 ± 0.002	0.957 ± 0.007	0.072	1000	
5.53 (M)	1.9 ± 0.1	2.1 ± 0.2	0.22	0.68 ± 0.02	1.8 ± 0.1	0.17	0.090 ± 0.002	1.17 ± 0.04	0.085	1050	
5.58 (A)	1.96 ± 0.04	1.04 ± 0.06	0.090	0.73 ± 0.04	0.9 ± 0.1	0.070				1050	
6.00 (T)	3.7 ± 0.2	2.2 ± 0.3	0.160							1140	53.8
6.03 (M)	3.7 ± 0.2	2.7 ± 0.5	0.16	1.55 ± 0.11	3.2 ± 0.4	0.20	0.187 ± 0.006	1.9 ± 0.3	0.071	1140	53.8
6.48 (B)	5.5 ± 0.2	4.6 ± 0.4	0.23							1180	55.0
6.56 (M)	5.9 ± 0.2	5.5 ± 0.6	0.26	2.71 ± 0.14	7.9 ± 1.2	0.60	0.324 ± 0.004	5.0 ± 0.2	0.145	1180	55.0
6.94 (P)	9.2 ± 0.4	21 ± 3	0.84	2.97 ± 0.12	17 ± 2	1.05	0.43 ± 0.02	15 ± 1.5	0.38	1240	57.3
7.02 (T)	8.9 ± 0.8	21 ± 5	0.79							1240	57.3
7.46 (P)	9.5 ± 0.8	63 ± 11	2.20	1.46 ± 0.08	26 ± 4	1.6	0.169 ± 0.006	17 ± 2	0.76		58.7
7.97 (B)	7.5 ± 0.7	160 ± 20	7.10	0.73 ± 0.04	50 ± 7	3.3					59.2
8.46 (B)	6.4 ± 0.7	390 ± 50	12.5	0.42 ± 0.02	110 ± 10	5.2					57.7
9.00 (S)	0.0050^{b}		19.0	0.0010^{b}		5.0					57.0

^a Data were obtained at 25 °C in 0.10 M buffer containing NaCl (to I = 0.10 M). Letters in parentheses denote buffer: F = formic acid, A = acetic acid, M = MES, T = Bis-Tris, B = BICINE, P = MOPS, and S = AMPSO. Represents k_{cat}/K_m rather than k_{cat}

Table 4: Intrinsic Steady-State Kinetic Parameters and Apparent pK_a 's for the Hydrolysis of cUMP by Wild-Type RNase A, D121N RNase A, and D121A RNase A^a

enzyme	pK_a	pK_b	pK_c	pK_d	$\begin{array}{c} k_{\rm cat}/K_{\rm m}^{\rm int}\\ ({\rm mM}^{-1}~{\rm s}^{-1}) \end{array}$
wild-type	5.58 ± 0.20	6.01 ± 0.17	3.80 ± 0.12		4.9 ± 1.8
D121N	5.48 ± 0.13	6.01 ± 0.06	3.80 ± 0.17	9.09 ± 0.17	1.60 ± 0.27
D121A	5.68 ± 0.97	5.84 ± 0.96			0.45 ± 0.95

enzyme	$pK_{a,s}$	$pK_{b,s}$	$pK_{c,s}$	$k_{\rm cat}^{\rm int1}$ (s ⁻¹)	$k_{\rm cat}^{\rm int2}$ (s ⁻¹)
wild-type	6.47 ± 0.14	8.63 ± 0.16	4.99 ± 0.15	10.5 ± 1.0	49 ± 14
D121N	6.32 ± 0.28	7.44 ± 0.15	4.52 ± 0.29	3.9 ± 1.1	24 ± 15
D121A	6.66 ± 0.22	6.95 ± 0.21	4.62 ± 0.75	1.0 ± 0.3	6.5 ± 3.2

enzyme	$pK_{a,p}$	$pK_{b,p}$	$K_{\rm p}^{\rm int}$ (mM)
wild-type	6.12 ± 0.10	8.53 ± 0.05	0.023 ± 0.002
D121N	6.03 ± 0.12	7.60 ± 0.10	0.020 ± 0.002
D121A	6.00 ± 0.09	7.83 ± 0.04	0.013 ± 0.001

^a Values were calculated by fitting the data in Table 3 to eqs 1−3.

position regardless of the crystallization conditions, solvent molecules, or ligand in the active site (62; L. W. Schultz and R. T. Raines, unpublished results). His12 forms a hydrogen bond between $N_{\delta 1}$ of its side chain and the main chain carbonyl of Thr45 and between $N_{\epsilon 2}$ and a water molecule or a ligand in the active site. The position of His12 in the D121N or D121A variants has not changed, overlapping quite well with His12 from the sD121N and sD121A structures. In the D121N structure, $N_{\epsilon 2}$ of His12 forms a 2.7 Å hydrogen bond with a water molecule in the active site. A similar 2.7 Å hydrogen bond is also found in the wild-type monoclinic structure. In the structure of the sD121N analogue, $N_{\epsilon 2}$ of His12 forms a 2.5 Å bond with a sulfate oxygen in the active site. In the D121A structure, N₆₂ of His12 forms a 2.7 Å hydrogen bond with an acetate oxygen in the active site. This acetate oxygen is located in the same position as the sulfate oxygen that forms a 2.8 Å hydrogen bond with $N_{\epsilon 2}$ in the wild-type trigonal structure. As in the sD121N analogue, His12 of the sD121A analogue forms a 2.5 Å hydrogen bond with a sulfate oxygen in the active site.

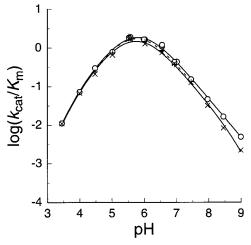


FIGURE 4: Plot of $log(k_{cat}/K_m)$ vs pH for the hydrolysis of cUMP by wild-type RNase A (O), D121N RNase A (x), and D121A RNase A (+). The curves are nonlinear least-squares fits of the data to eq 1. To simplify comparisons, the values of $k_{\text{cat}}/K_{\text{m}}$ for D121N RNase A and D121A RNase A were multiplied by scaling factors of 2.2 and 13, respectively.

Lys41. The side chain of Lys41 has been implicated in transition-state stablization, contributing 10⁵-fold to catalysis (63). In addition, N_{ξ} of Lys41 has been shown to form a hydrogen bond with the 2'-oxygen of 3'-CMP and to a bridging vanadyl oxygen of uridine 2',3'-cyclic vanadate (49, 58). In the D121N and D121A variants, the side chain of Lys41 has clearly defined density and exists in a fully extended conformation. In the wild-type monoclinic and trigonal structures, N_{ξ} of Lys41 forms a hydrogen bond (2.6 and 2.7 Å, respectively) with $O_{\delta 1}$ of Asn44. The structures of the D121N and D121A variants have the same hydrogen bond (2.8 and 2.9 Å, respectively) between N_{ξ} of Lys41 and $O_{\delta 1}$ of Asn44, but N_{\gamma\$} in the D121A enzyme makes an additional 2.5 Å hydrogen bond with a new water molecule in the active site. In the sD121N and sD121A analogues, N_{ξ} of Lys41 also forms a hydrogen bond (2.9 and 3.2 Å, respectively) to $O_{\delta 1}$ of Asn44. As in the D121A variant, however, N_{ξ} of Lys41 in the sD121N analogue forms an additional 2.7 Å hydrogen bond with a water molecule in

FIGURE 5: Plot of $\log(k_{\text{cat}})$ vs pH for the hydrolysis of cUMP by wild-type RNase A (\bigcirc), D121N RNase A (\times), and D121A RNase A (+). The curves are nonlinear least-squares fits of the data to eq 2. To simplify comparisons, the values of k_{cat} for D121N RNase A and D121A RNase A were multiplied by scaling factors of 1.6 and 16, respectively.

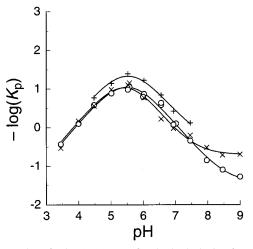


FIGURE 6: Plot of $-\log(K_p)$ vs pH for the hydrolysis of cUMP by wild-type RNase A (\bigcirc), D121N RNase A (\times), and D121A RNase A (+). The curves are nonlinear least-squares fits of the data to eq 3. No scaling factors were applied to these data.

the active site. In the sD121A analogue, N_{ξ} also forms a 2.8 Å hydrogen bond with $O_{\epsilon 1}$ of Gln11.

Residues 65–72. Residues 65–72 form a loop between $\alpha 3$ and $\beta 2$. This loop resides near the active site and contains Lys66, which is implicated in the binding of a substrate phosphoryl group (59). Lys66 makes two interactions with Asp121 in the wild-type monoclinic RNase A structure: a 2.8 Å hydrogen bond from the main chain N of Lys66 to $O_{\delta 2}$ of Asp121 and a 3.0 Å hydrogen bond from N_{ζ} of Lys66 to the main chain O of Asp121. In the wild-type trigonal structure, the main chain N of Lys66 also forms a 3.3 Å hydrogen bond to $O_{\delta 2}$ of Asp121, but N_{ζ} of Lys66 forms no interactions with the protein.

The hydrogen bond between the main chain N of Lys66 and O₆₂ of Asp121 restricts conformational entropy by linking two parts of the native enzyme that are proximal in space but distal in sequence. ¹H NMR data suggest that this hydrogen bond is especially strong (64). In X-ray diffraction analyses, the length of this hydrogen bond appears to depend on the crystallization conditions, being 2.8–2.9 Å in salt-free crystals and 2.9–3.3 Å in crystals containing salt.

Movement of the loop containing residues 65–72 away from the active site has been proposed to be a major contributor to the reduced activity of the sD121N and sD121A analogues (30). The main chain atoms of residues 65–72 in the trigonal sD121N structure have an average shift of 0.76 Å away from the wild-type main chain. In contrast, the same atoms in the monoclinic D121N structure have an average shift of 0.45 Å away from the wild-type main chain, suggesting that the crystallization conditions and not the amino acid substitution are responsible for the larger shift.

Other significant differences exist between the D121N and sD121N structures (Figure 3A). The hydrogen bond between the main chain N of Lys66 and $O_{\delta 1}$ of Asn121 is 2.9 Å in the D121N variant but 3.2 Å in the sD121N analogue. The side chain N_{ξ} of Lys66 forms a 3.3 Å hydrogen bond with the main chain O of Asn121 in the D121N variant but a 3.5 Å hydrogen bond with $O_{\delta 1}$ of Asn121 in the sD121N analogue. Semisynthetic F120L and F120Y analogues of RNase A in the trigonal form also form hydrogen bonds between N_{ζ} of Lys66 and $O_{\delta 1}$ of Asp121, but neither forms a hydrogen bond with the main chain O of Asp121 (65). Another difference is more dramatic, as it involves the flipping of the Lys66-Asn67 peptide bond. The ϕ, ψ angles of Lys66 and Asn67 for wild-type RNase A (-56°,-35° and $-85^{\circ},1^{\circ}$) and the D121N variant $(-55^{\circ},-39^{\circ})$ and $-82^{\circ}, -5^{\circ}$) are similar, but those for the sD121N analogue $(-84^{\circ},83^{\circ})$ and $166^{\circ},-26^{\circ}$ differ. In the sD121N structure, a water molecule occupies the position of the main chain O in the D121N structure. This water molecule in the sD121N structure is only 2.1 Å (which is an unrealistic distance) from the main chain N of Asn67. We therefore suspect that the discrepancy between the location of residues 65-72 in the D121N and sD121N structures is largely due to an error in electron density map interpretation.

Residues 65-72 in D121A RNase A are almost identical in structure to residues 65-72 in the sD121A analogue. The loss of the hydrogen bond between the side chain of residue 121 and the main chain N of Lys66 has caused the main chain atoms in residues 65-72 in both structures to move away from the wild-type main chain by an average of 0.7 Å. Moreover, the Lys66 side chain forms no hydrogen bonds to the protein in either structure. The absence of this hydrogen bond is, however, not uncommon [PDB entries 1lsq and 1rnc (66, 67)]. Although the Asp121 side chain to Lys66 main chain hydrogen bond appears to be important in stabilizing the structure of residues 65-72, it is unlikely that hydrogen bonds to the side chain of Lys66 play a significant role in that stabilization.

His119. In the classical mechanism of RNase A catalysis (Figure 1), His119 acts as an acid by donating a proton to the 5"-oxygen to facilitate its displacement (33, 61). The side chain of His119 has been shown to occupy two distinct positions, denoted as A and B, which are related by a rotation around $χ_1$ to dihedral angles of approximately 159 and -60° , respectively (68, 69). The active-site histidine residue in carbonic anhydrase II undergoes a similar rotation (70). There has been considerable debate as to which position of His119 in RNase A is catalytically relevant because the imidazole nitrogens have access to the active site in both orientations (30). It has also been argued that His119 is in position A during catalysis of transphosphorylation and in position B during catalysis of hydrolysis (69). In position A, $N_{ε2}$ of

The side chain of His119 occupies position A in the D121N structure and overlaps with His119 in the wild-type structure almost exactly, with a χ_1 of 160°. The side chain $N_{\epsilon 2}$ of His119 forms a 2.6 Å hydrogen bond with $N_{\delta 2}$ of Asn121, and $N_{\delta 1}$ forms no hydrogen bonds. Similarly, the side chain of His119 occupies position A in the D121A structure but has shifted toward the active site by a 15° rotation about a χ_1 of 175°. In the absence of a hydrogenbonding side chain at residue 121, $N_{\epsilon 2}$ of His119 forms no hydrogen bonds, and $N_{\delta 1}$ of His119 forms a 2.5 Å hydrogen bond with the acetate ion in the active site. In the sD121A $(\chi_1 = -50^\circ)$ and sD121N $(\chi_1 = -49^\circ)$ analogues, the side chain of His119 occupies position B. Others have concluded that, in the absence of a hydrogen bond with the side chain of Asp121, His119 would prefer position B over position A (30). We too expected a priori that replacement of Asp121 would shift the equilibrium in favor of position B because of the loss of the favorable His...Asp interaction. Yet in the structures of crystalline D121N RNase A and D121A RNase A, His119 is in position A (Figures 2 and 3). It appears that the position of His119 is more dependent on crystallization conditions than on the identity of residue 121. Similarly, the position of His119 has been shown to depend on the pH (65) and ionic strength (62) of the crystallization solution. Finally, unlike in serine hydrolases (71), $C_{\epsilon 1}$ of His119 is not proximal to a main chain O in either position A or position B. (Likewise, $C_{\epsilon 1}$ of His12 is not near a main chain O.)

Asn121. At a resolution of 1.6 Å, it is impossible to distinguish $N_{\delta 2}$ and $O_{\delta 1}$ in the side chain of Asn121. It is, however, possible to distinguish these atoms on the basis of their hydrogen bonding schemes. In our model, $O_{\delta 1}$ of Asn121 occupies a position similar to that of $O_{\delta 2}$ in wildtype RNase A, forming a 2.9 Å hydrogen bond with the main chain N of Lys66. This orientation is sensical because it is unlikely that $N_{\delta 2}$ of Asn121 could be close to a main chain nitrogen. This orientation places $N_{\delta 2}$ of Asn121 near $N_{\epsilon 2}$ of His119. At pH 5.3 (which is the pH of the crystallization buffer), $N_{\epsilon 2}$ of His119 should be protonated, thereby rendering its interaction with the Asn121 side chain unfavorable. But, the crystals of D121N RNase A were grown in MPD, a solvent less polar than water. Accordingly, the pK_a of His119 in the crystal is likely to be smaller than in water. We therefore propose that His119 is in the neutral imidazole state in which $N_{\delta 1}$ is protonated and that $N_{\epsilon 2}$ forms a 2.6 Å hydrogen bond with the side chain of Asn121.

A water molecule forms a bridging hydrogen bond between $N_{\delta 2}$ of Asn121 and the main chain N of Asn67. The $N_{\delta 2}$ ···water hydrogen bond distance is 3.1 Å with an $N_{\delta 2}$ -H-O angle of 120°. This water molecule, which is conserved in several RNase A structures, could stabilize the conformation of residues 65–72 by linking them to residue 121. In the sD121N analogue, the water molecule has moved significantly, forming a long hydrogen bond with the main chain N of Asn67 and perhaps resulting in a shift of residues 65–72 away from the active site. Still, the side chains of Asn121 in the D121N and sD121N structures overlap almost exactly (Figure 3A).

Ala121. The main chain near residue 121 in the D121A variant remains in the same conformation as in the trigonal wild-type structure. The side chain C_{β} of Ala121 overlaps directly with C_{β} of Asp121 in the wild-type RNase structure. In the D121A variant, a water molecule takes the place of $O_{\delta 2}$ of Asp121 and forms a 3.2 Å hydrogen bond with the main chain N of Lys66. This water molecule is absent from the sD121A structure. In addition, the conserved water molecule that forms a bridging hydrogen bond between $O_{\delta 1}$ of Asp121 and the main chain N of Asn67 in the wild-type enzyme is missing in the D121A and sD121A structures.

Catalysis by D121N RNase A and D121A RNase A. The effect of replacing Asp121 on $k_{\rm cat}/K_{\rm m}$ and $k_{\rm cat}$ is small, but significant (Tables 2–4). Substitution with an asparagine residue decreases the rate constants for transphosphorylation by 10-fold and those for hydrolysis by $10^{0.5}$ -fold (Table 2). Substitution with an alanine residue decreases the rate constants for transphosphorylation by 10^2 -fold and those for hydrolysis by 10-fold (Tables 2–4). The differential effects on transphosphorylation and hydrolysis are still greater than these values suggest because a chemical step does not limit the rate of cleavage of UpA or poly(C) (72). We conclude that the contribution of Asp121 to transphosphorylation is greater than that to hydrolysis.

The small contribution of Asp121 to catalysis belies the importance of the hydrogen bond between His119 and Asp121. This contribution is similar to that observed in an analogous study on the structure and function of phosphatidylinositol-specific phospholipase C (73). Yet, a His···Asp dyad has been proposed to form a low-barrier hydrogen bond of extraordinary strength during catalysis by the protease chymotrypsin (74, 75), though this interpretation is controversial (76). One criterion for such a bond is an ¹H chemical shift of 17–20 ppm (77). The ${}^{1}H$ chemical shift of $N_{e2}H$ of His119 appears at a much higher field, <12 ppm (J. L. Markley, personal communication). By this criterion, the His···Asp dyad of unliganded RNase A does not have a lowbarrier hydrogen bond. In theory, the His···Asn dyad of D121N RNase A could also form a low-barrier hydrogen bond. Another criterion for such a bond is that it arise from acids with matched pK_a values (79). The pK_a values of imidazole [14.2 (81)] and acetamide [15.1 (80)], which are reasonable models for the side chains of His119 and Asn121, respectively, are indeed similar. The existence of a lowbarrier hydrogen bond would, however, require that the side chain of His119 be in its neutral imidazole form. pH-rate profiles for catalysis by D121N RNase A (Figures 4–6) and ¹H NMR titrations of His119 in the D121N variant (54) indicate that His119 titrates between its imidazole and imidazolium forms with a p K_a value similar to that of His119 in wild-type RNase A. These data deny the existence of a low-barrier hydrogen bond in the His···Asn dyad.

The kinetics of catalysis by D121N RNase A and D121A RNase A illuminate mechanistic aspects of the equilibrium of His119. A Coulombic interaction, like that between a cationic histidine side chain and an anionic aspartate side chain, has a long range. In contrast, a hydrogen bond, like that between a histidine side chain and an aspartate or asparagine side chain, has a short range. Dispersion interactions between a histidine side chain and an alanine side chain are insignificant at a distance beyond van der Waals radii. We find that RNase A variants with an aspartate, asparagine, or alanine residue in position 121 have different abilities to catalyze the hydrolysis reaction (Tables 3 and 4). We therefore conclude that His119 is likely to be proximal to residue 121 during catalysis of hydrolysis and that hydrolysis can occur with His119 in position A.

pH Dependence of Catalysis. The overall shape of the pH-rate profiles for catalysis by RNase A can be interpreted by using a model in which the only important protonation states are those of the two active-site histidine residues, which act in concert during catalysis [Figure 1 (82)]. Accordingly, log-log plots of the pH-rate profiles for k_{cat}/K_{m} and k_{cat} are expected to be bell-shaped with slopes approaching 1. The $k_{\text{cat}}/K_{\text{m}}$ profiles that have been reported are indeed bellshaped, but the shape of the k_{cat} profiles varies with the substrate (83-86). In general, workers who have used cCMP as the substrate have found that the slope is 1 for the acidic leg of the k_{cat} profiles, but those who have used cUMP have found that the slope is ≤ 1 . The large values of $K_{\rm m}$ and strong product inhibition observed for the hydrolysis step at high pH make the basic leg difficult to characterize. These factors are likely to be the cause of the discrepant values reported for k_{cat} at high pH (83).

Our data on the pH dependence of $k_{\rm cat}/K_{\rm m}$ for the hydrolysis of cUMP are consistent with those of previous studies (for a review, see ref 87). Our data on the pH dependence of $k_{\rm cat}$ for the hydrolysis of cUMP are most similar to those of workers who, like us, accounted for product inhibition by using the integrated rate equation (86). This approach yields a slope near 1 at values of pH <4.5, as well as values of $k_{\rm cat}$ and $K_{\rm m}$ at values of pH >7 higher than those reported by others (84, 85).

Macroscopic pK_a values for the two histidine residues of wild-type RNase A can be derived from the k_{cat}/K_m profile (Table 4). The values are consistent with results from NMR spectroscopy (54, 88–91). Differences for any given pK_a value are within error, especially when recalling that, when two pK_a 's have similar values, the average value is much better determined from pH-rate profiles than are the individual values (92). The pK_a values of the D121N and D121A enzymes that are derived from pH-rate profiles (Table 4) do not deviate significantly from those of wild-type RNase A, at least when I = 0.10 M.

Role of Asp121 in Catalysis. pH—rate profiles for catalysis by D121N RNase A are similar to those for catalysis by the wild-type enzyme (Figures 4–6). This similarity suggests that the same titratable groups are participating in catalysis and that these groups have similar pK_a values. Yet in the structure of D121N RNase A, $N_{\delta 2}$ rather than $O_{\delta 1}$ of Asn121 is oriented toward His119. This orientation appears to stabilize a neutral form of the imidazolyl side chain in which

 $N_{\rm e2}$ is unprotonated and $N_{\delta 1}$ is protonated. This tautomer of His119 is likely to be ineffective, either as an acid during the transphosphorylation reaction or as a base during the hydrolysis reaction. The major role of Asp121 thus appears to be to orient the proper tautomer of His119 for catalysis. This role is similar to that assigned to the aspartate residue in the catalytic triad of trypsin (26). In the D121A enzyme, Ala121 cannot interact with the imidazolyl group of His119. Yet, D121N RNase A is a better catalyst than is D121A RNase A (Tables 2–4). Thus, the His119···Asn121 interaction observed in the crystalline D121N enzyme (Figures 2 and 3) is unlikely to exist in solution. Rather, Asn121 is likely to move such that His119 can effect catalysis.

Conclusions. Replacing Asp121, a residue that is conserved in pancreatic ribonucleases, with an asparagine or alanine residue has no effect on the overall three-dimensional structure of RNase A. Replacing Asp121 decreases the rate of transphosphorylation by $\leq 10^2$ -fold and the rate of hydrolysis by ≤ 10 -fold. Thus, Asp121 of RNase A makes a significant but not a substantial contribution to catalysis. The likely role of Asp121 is simply to position the proper tautomer of His119, which Asn121 in the crystalline D121N enzyme fails to do. We speculate that the role of Asp121 in catalysis alone is unlikely to warrant its complete conservation in pancreatic ribonucleases (54).

ACKNOWLEDGMENT

We thank B. R. Kelemen, P. A. Leland, and B. M. Templer for comments on the manuscript and Prof. J. L. Markley for advice. We are grateful to Prof. I. Rayment, Prof. H. M. Holden, and the members of their research groups for many helpful conversations and for the use of their X-ray data collection and computational facilities.

REFERENCES

- 1. Neurath, H. (1989) in *Proteolytic Enzymes: A Practical Approach* (Beynon, R. J., and Bond, J. S., Eds.) pp 1–13, IRL Press, New York.
- 2. Blow, D. M., Birktoft, J. J., and Hartley, B. S. (1969) *Nature* 221, 337–340.
- 3. Blow, D. M. (1997) Trends Biochem. Sci. 22, 405-408.
- 4. Neurath, H. (1984) Science 224, 350-357.
- 5. Brady, L., Brzozowski, A. M., Derewenda, Z. S., Dodson, E., Dodson, G., Tolley, S., Turkenburg, J. P., Christiansen, L., Huge-Jensen, B., Norskov, L., Thim, L., and Menge, U. (1990) *Nature* 343, 767–770.
- Winkler, F. K., D'Arcy, A., and Hunziker, W. (1990) Nature 343, 771–774.
- 7. Martinez, C., De Geus, P., Lauwereys, M., Matthyssens, G., and Cambillau, C. (1992) *Nature 356*, 615–618.
- 8. Choi, H.-K., Tong, L., Minor, W., Dumas, P., Boege, U., Rossmann, M. G., and Wengler, G. (1991) *Nature 354*, 37–43.
- Faustinella, F., Chang, A., Van Biervliet, J. P., Rosseneu, M., Vinaimont, N., Smith, L. C., Chen, S.-H., and Chan, L. (1991) J. Biol. Chem. 266, 14418–14424.
- Liao, D.-I., Breddam, K., Sweet, R. M., Bullock, T., and Remington, S. J. (1992) *Biochemistry* 31, 9796–9812.
- Sussman, J. L., Harel, M., Frolow, F., Oefner, C., Goldman, A., Toker, L., and Silman, I. (1991) Science 253, 872–879.
- Li, Y., and Tsai, M.-D. (1993) J. Am. Chem. Soc. 115, 8523

 8526.
- 13. Verschueren, K. H. G., Seljée, F., Rozeboom, H. J., Kalk, K. H., and Dijkstra, B. W. (1993) *Nature 363*, 693–698.

- 14. Pathak, D., and Ollis, D. (1990) J. Mol. Biol. 214, 497-525.
- Christianson, D. W., and Alexander, R. S. (1989) J. Am. Chem. Soc. 111, 6412–6419.
- 16. Beintema, J. J. (1987) Life Chem. Rep. 4, 333-389.
- Beintema, J. J., Schüller, C., Irie, M., and Carsana, A. (1988)
 Prog. Biophys. Mol. Biol. 51, 165–192.
- 18. Gandour, R. D. (1981) Bioorg. Chem. 10, 169-176.
- Zimmerman, S. C., and Cramer, K. D. (1988) J. Am. Chem. Soc. 110, 5906-5908.
- Cramer, K. D., and Zimmerman, S. C. (1990) J. Am. Chem. Soc. 112, 3680–3682.
- Schowen, R. L. (1988) in *Mechanistic Principles of Enzyme Activity* (Liebman, J. F., and Greenberg, A., Eds.) pp 119–168, VCH Publishers, New York.
- Umeyama, H., Nakagawa, S., and Fujii, T. (1979) Chem. Pharm. Bull. 27, 974–980.
- Craik, C. S., Largman, C., Fletcher, T., Roczniak, S., Barr, P. J., Fletterick, R., and Rutter, W. J. (1985) Science 228, 291–297.
- Corey, D. R., and Craik, C. S. (1992) J. Am. Chem. Soc. 114, 1784–1790.
- 25. Carter, P., and Wells, J. A. (1988) *Nature 332*, 564–568.
- Sprang, S., Standing, T., Fletterick, R. J., Stroud, R. M., Finer-Moore, J. F., Xuong, N. H., Hamlin, R., Rutter, W. J., and Craik, C. S. (1987) *Science* 237, 905–909.
- 27. Raines, R. T. (1998) Chem. Rev. 98, 1045-1065.
- 28. Stern, M. S., and Doscher, M. S. (1984) FEBS Lett. 171, 253–256
- Cederholm, M. T., Stuckey, J. A., Doscher, M. S., and Lee,
 L. (1991) Proc. Natl. Acad. Sci. U.S.A. 88, 8116–8120.
- de Mel, V. S. J., Martin, P. D., Doscher, M. S., and Edwards, B. F. P. (1992) *J. Biol. Chem* 267, 247–256.
- Trautwein, K., Holliger, P., Stackhouse, J., and Benner, S. A. (1991) FEBS Lett. 281, 275–277.
- Harper, J. W., and Vallee, B. L. (1988) Proc. Natl. Acad. Sci. U.S.A. 88, 7139-7143.
- 33. Findlay, D., Herries, D. G., Mathias, A. P., Rabin, B. R., and Ross, C. A. (1961) *Nature 190*, 781–784.
- 34. Cuchillo, C. M., Parés, X., Guasch, A., Barman, T., Travers, F., and Nogués, M. V. (1993) *FEBS Lett.* 333, 207–210.
- 35. Thompson, J. E., Venegas, F. D., and Raines, R. T. (1994) *Biochemistry 33*, 7408–7414.
- 36. Doolittle, R. F. (1992) Protein Sci. 1, 191-200.
- 37. Studier, F. W., and Moffatt, B. A. (1986) *J. Mol. Biol. 189*, 113–130
- Ogilvie, K. K., Beaucage, S. L., Schifman, A. L., Theriault, N. Y., and Sadana, K. L. (1978) *Can. J. Chem.* 56, 2768– 2780.
- Beaucage, S. L., and Caruthers, M. H. (1981) Tetrahedron Lett. 22, 1859–1862.
- 40. Tartof, K. D., and Hobbs, C. A. (1987) Bethesda Res. Lab. Focus 9, 12.
- del Cardayré, S. B., Ribó, M., Yokel, E. M., Quirk, D. J., Rutter, W. J., and Raines, R. T. (1995) *Protein Eng.* 8, 261–273.
- 42. Kunkel, T. A., Roberts, J. D., and Zakour, R. A. (1987) *Methods Enzymol.* 154, 367–382.
- Sela, M., Anfinsen, C. B., and Harrington, W. F. (1957) *Biochim. Biophys. Acta* 26, 502–512.
- 44. Kabsch, W. (1988) J. Appl. Crystallogr. 21, 67-71.
- 45. Kabsch, W. (1988) J. Appl. Crystallogr. 21, 916-924.
- 46. Tronrud, D. E., Ten-Eyck, L. F., and Matthews, B. W. (1987) *Acta Crystallogr., Sect. A* 43, 489–501.
- 47. Jones, T. A. (1985) Methods Enzymol. 115, 157-171.
- 48. Wlodawer, A., Anders, L. A., Sjölin, L., and Gilliland, G. L. (1988) *Biochemistry* 27, 2705–2717.
- 49. Zegers, I., Maes, D., Thi, M. H., Poortmans, F., Palmer, R. A., and Wyns, L. (1994) *Protein Sci. 3*, 2322–2339.
- Warshaw, M. M., and Tinoco, I. (1966) J. Mol. Biol. 20, 29–38.
- Yakovlev, G. I., Moiseyev, G. P., Bezborodova, S. I., Both,
 V., and Sevcik, J. (1992) Eur. J. Biochem. 204, 187–190.
- Segel, I. H. (1975) Enzyme Kinetics, John Wiley & Sons, New York.

- Cornish-Bowden, A. (1995) Fundamentals of Enzyme Kinetics, Portland Press, London.
- Quirk, D. J. (1996) Ph.D. Thesis, University of Wisconsin— Madison, Madison, WI.
- 55. Tanokura, M. (1983) J. Biochem. 94, 1621-1630.
- Kartha, G., Bello, J., and Harker, D. (1967) Nature 213, 862

 865.
- 57. Gilliland, G. L. (1997) in *Ribonucleases: Structures and Functions* (D'Alessio, G., and Riordan, J. F., Eds.) pp 306–341, Academic Press, New York.
- Wlodawer, A., Miller, M., and Sjölin, L. (1983) Proc. Natl. Acad. Sci. U.S.A. 80, 3628-3631.
- Parés, X., Nogués, M. V., de Llorens, R., and Cuchillo, C. M. (1991) Essays Biochem. 26, 89–103.
- Ladner, J. E., Wladkowski, B. D., Svensson, L. A., Sjölin, L., and Gilliland, G. L. (1997) *Acta Crystallogr., Sect. D* 53, 290– 301.
- Thompson, J. E., and Raines, R. T. (1994) J. Am. Chem. Soc. 116, 5467-5468.
- Fedorov, A. A., Joseph-McCarthy, D., Fedorov, E., Sirakova, D., Graf, I., and Almo, S. C. (1996) *Biochemistry 35*, 15962– 15979.
- 63. Messmore, J. M., Fuchs, D. N., and Raines, R. T. (1995) *J. Am. Chem. Soc.* 117, 8057–8060.
- 64. Baker, W. R., and Kintanar, A. (1996) *Arch. Biochem. Biophys.* 327, 189–199.
- de Mel, V. S. J., Doscher, M. S., Martin, P. D., and Edwards, B. F. P. (1994) FEBS Lett. 349, 155–160.
- 66. Capasso, S., Giordano, F., Mattia, C. A., Mazzarella, L., and Zagari, A. (1983) *Biopolymers* 22, 327–332.
- 67. Aguilar, C. F., Thomas, P. J., Mills, A., Moss, D. S., and Palmer, R. A. (1992) *J. Mol. Biol. 224*, 265–267.
- 68. Borkakoti, N., Moss, D. S., and Palmer, R. A. (1982) *Acta Crystallogr.*, Sect. B 38, 2210–2217.
- 69. Borkakoti, N. (1983) Eur. J. Biochem. 132, 89-94.
- 70. Nair, S. K., and Christianson, D. W. (1991) *J. Am. Chem. Soc.* 113, 9455–9458.
- 71. Derewenda, Z. S., Derewenda, U., and Kobos, P. M. (1994) *J. Mol. Biol.* 241, 83–93.
- Thompson, J. E., Kutateladze, T. G., Schuster, M. C., Venegas, F. D., Messmore, J. M., and Raines, R. T. (1995) *Bioorg. Chem.* 23, 471–481.
- 73. Gässler, C. S., Ryan, M., Liu, T., Griffith, O. H., and Heinz, D. W. (1997) *Biochemistry 36*, 12802–12813.
- Frey, P. A., Whitt, S. A., and Tobin, J. B. (1994) Science 264, 1927–1930.
- Cassidy, C. S., Lin, J., and Frey, P. A. (1997) *Biochemistry* 36, 4576–4584.
- Ash, E. L., Sudmeier, J. L., De Fabo, E. C., and Bachovchin, W. W. (1997) Science 278, 1128–1132.
- Cleland, W. W., and Kreevoy, M. M. (1994) Science 264, 1887–1890.
- Patel, D. J., Woodward, C. K., and Bovey, F. A. (1972) Proc. Natl. Acad. Sci. U.S.A. 69, 599

 –602.
- 79. Cleland, W. W. (1992) Biochemistry 31, 317-319.
- 80. Bordwell, F. G. (1988) Acc. Chem. Res. 21, 456-463.
- 81. Walba, H., and Isensee, R. W. (1955) *J. Am. Chem. Soc.* 77, 5488-5492.
- 82. Sowa, G. A., Hengge, A. C., and Cleland, W. W. (1997) *J. Am. Chem. Soc.* 119, 2319–2320.
- 83. Herries, D. G., Mathias, A. P., and Rabin, B. R. (1962) *Biochem. J.* 85, 127–134.
- 84. delRosario, E. J., and Hammes, G. G. (1969) *Biochemistry 8*, 1884–1889.
- Machuga, E., and Klapper, M. H. (1977) *Biochim. Biophys. Acta* 481, 526–541.
- Eftink, M. R., and Biltonen, R. L. (1983) Biochemistry 22, 5123-5134.
- Richards, F. M., and Wyckoff, H. W. (1971) The Enzymes IV, 647–806.
- 88. Meadows, D. H., Roberts, G. C. K., and Jardetzky, O. (1969) J. Mol. Biol. 45, 491–511.

- 89. Roberts, G. C. K., Dennis, E. A., Meadows, D. H., Cohen, J. S., and Jardetzky, O. (1969) *Proc. Natl. Acad. Sci. U.S.A.* 62, 1151–1158.
- 90. Markley, J. L., and Finkenstadt, W. R. (1975) *Biochemistry* 14, 3562-3566.
- 91. Patel, D. J., Canuel, L. L., and Bovey, F. A. (1975) *Biopolymers* 14, 987–997.
- 92. Brocklehurst, K. (1994) Protein Eng. 7, 291-299.
- 93. Kraulis, P. J. (1991) *J. Appl. Crystallogr.* 24, 946–950. BI9727660